



Residue

Risk reduction of chemical residues in soils and crops:
impact due to wastewater used for irrigation


Research on the environmental fate of organic chemicals using ^{14}C -radiolabelling

Dr. Dieter Hennecke, Dr. Kerstin Derz,
Fraunhofer IME
International Spring School, May 2023

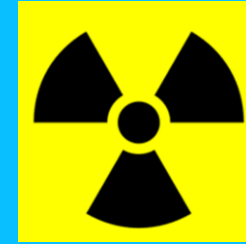
Carbon-14 / ^{14}C



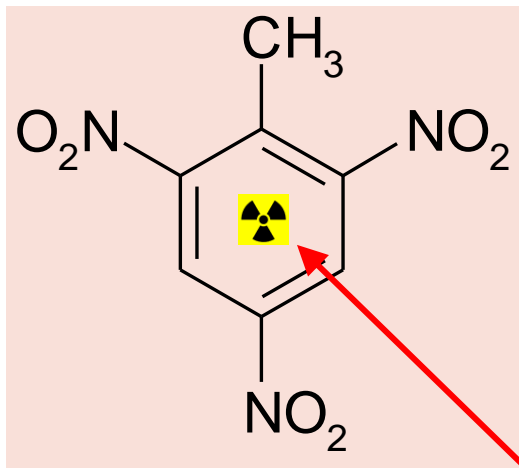
- Carbon-14, C-14, ^{14}C or radiocarbon
- Radioactive isotope
- Unstable with a half life of about 5700 years, ^{14}C decays into nitrogen-14 (^{14}N) through beta decay
- Sources: cosmic ray action on nitrogen in the atmosphere (cosmogenic nuclide), open air nuclear testing (1955-1980)

Carbon-12	Carbon-13	Carbon-14 
6 Electrons 6 Protons 6 Neutrons	6 Electrons 6 Protons 7 Neutrons	6 Electrons 6 Protons 8 Neutrons
Nuclear/mass number $6+6= 12$	Nuclear/mass number $6+7= 13$	Nuclear/mass number $6+8= 14$
99% of all carbon	up to 1% of all carbon	trace amounts (ca. 1 per 10^{12} atoms)

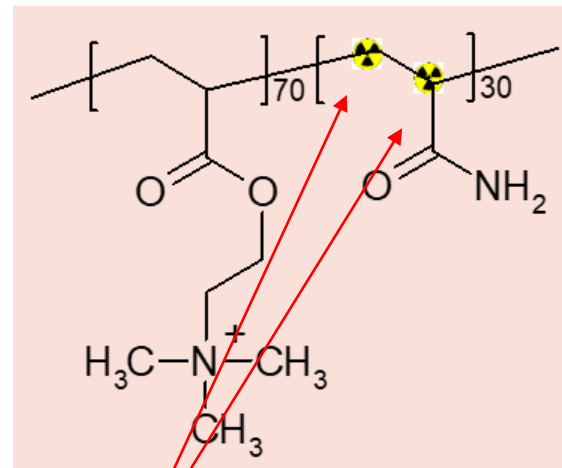
^{14}C -Labelling



2,4,6-TNT
(explosive)



Polyacrylamide
(polymere)



One or more ^{12}C -atoms replaced by ^{14}C -atoms

- Radioactivity
- Unit: Becquerel (**Bq**), an older unit is Curie (**Cu**)
- Labelling in the aromatic ring (U-label) or at specific positions in the molecule

^{14}C -Labelling



- Radiation protection: License for ^{14}C necessary
- Handling of ^{14}C -labelled test substances only in supervised or controlled areas (laboratories or outdoor facilities)

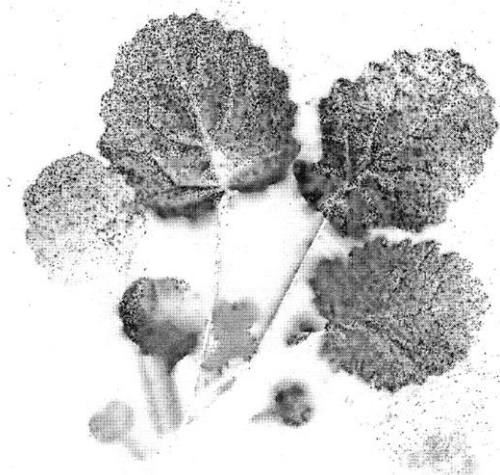


Measuring radioactivity



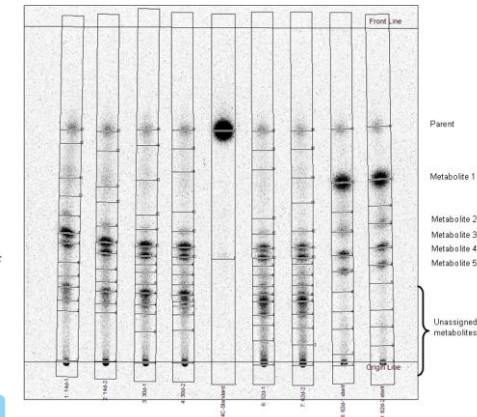
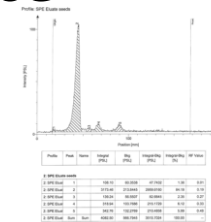
Detection of radioactivity (non-specific)

- Monitors measuring radioactivity on surfaces
- Liquid Scintillation Counting (LSC): Quantification of radioactivity in liquids
- Oxidizer: Combustion of solids followed by quantification of radioactivity by Liquid Scintillation Counting (LSC)



Chromatographic measurements – What kind of substance (still including the radiolabel) is it? (specific)

- Radio-HPLC: HPLC equipped with a radiodetector
- Radio-TLC: TLC plates exposed to imaging plates sensitive to radioactive radiation



Aim and advantages of radiolabelling



- Conventional analytical method (e.g. residue analytical method) for environmental matrices like soil, sediment, plant tissue etc.:

You need to know for which substance you are looking for!

- ^{14}C -labelling:

The substance is carrying a label so all metabolites until $^{14}\text{CO}_2$ can be tracked, i.e. the fate of the substance can be traced back by the radioactivity



Any radioactivity detected must be derived from the parent compound!

Where is it?

Is it parent compound?

Degradation to CO_2 ?

Are there metabolites detectable? Which ones?

Is there adsorption to soil or plant matrix?

Is something volatile formed?

Natural matrices like soil, plant material, sediment etc. are very complex and contain thousands of different organic substances.

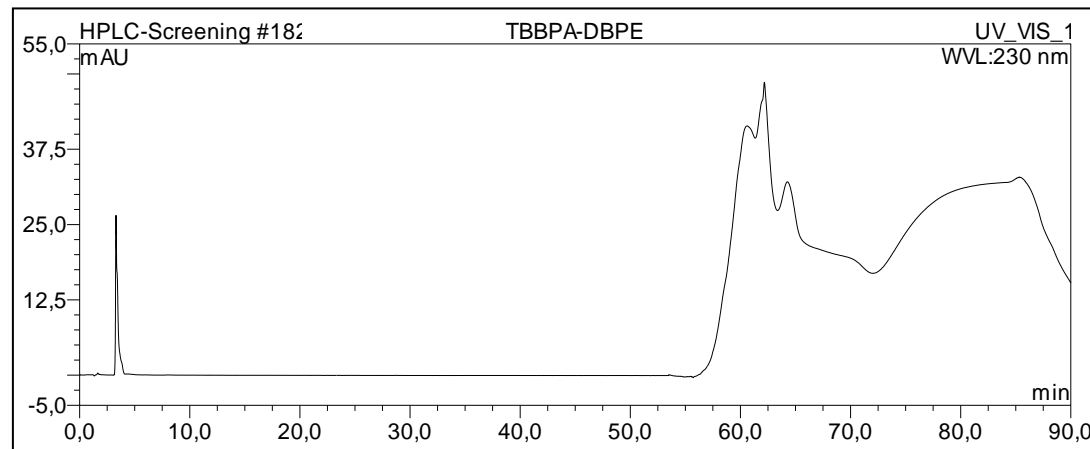
To find a definite substance is like looking for a needle in a haystack



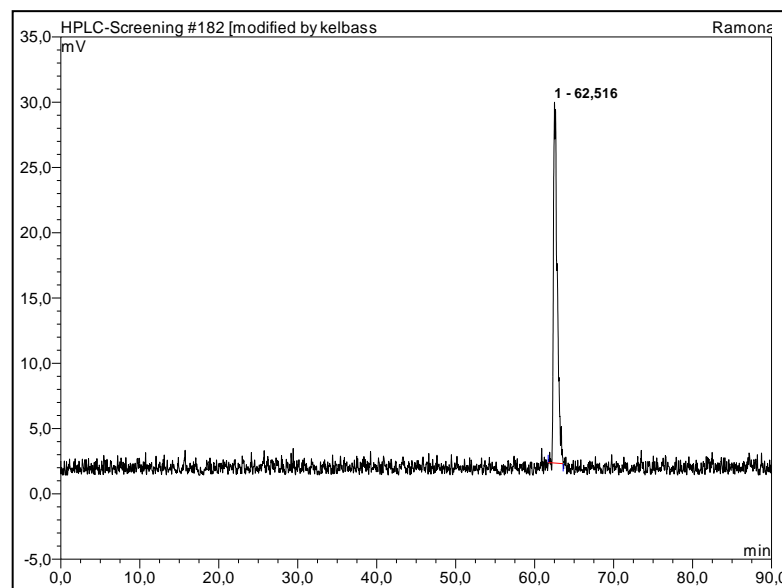
Aim and advantages of radiolabelling



HPLC analysis:
UV-detection



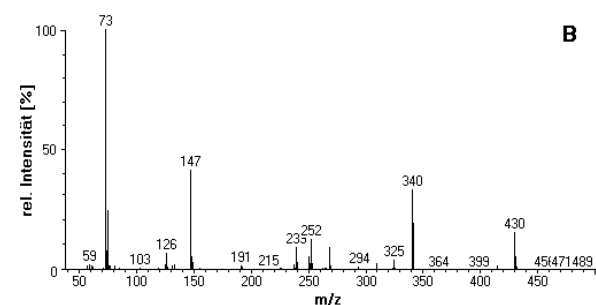
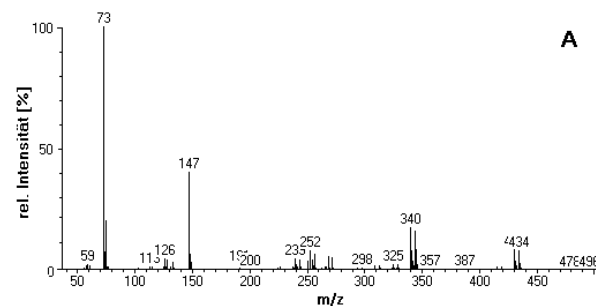
HPLC analysis:
radiodetection



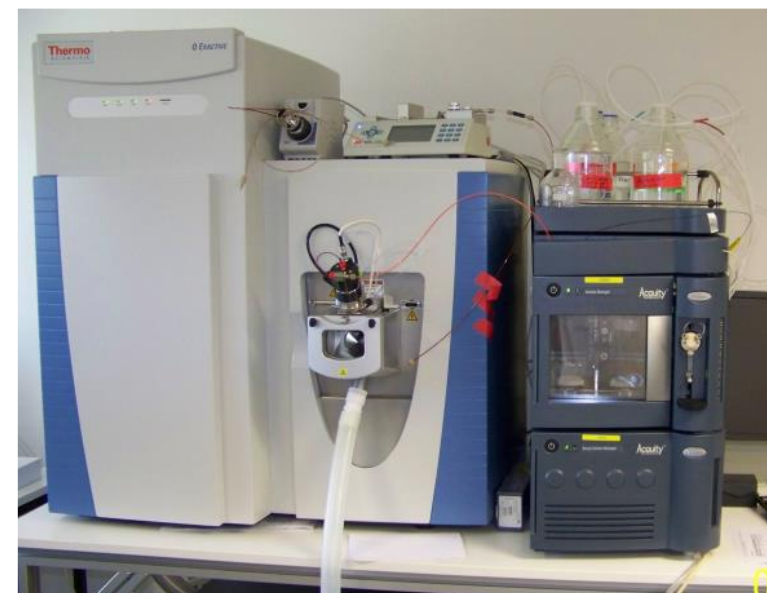
Aim and advantages of radiolabelling



- Metabolites can be traced back to the parent compound definitely
- Identification of the metabolites is possible by LC-HRMS and radiodetection and/or the characteristic exact masses



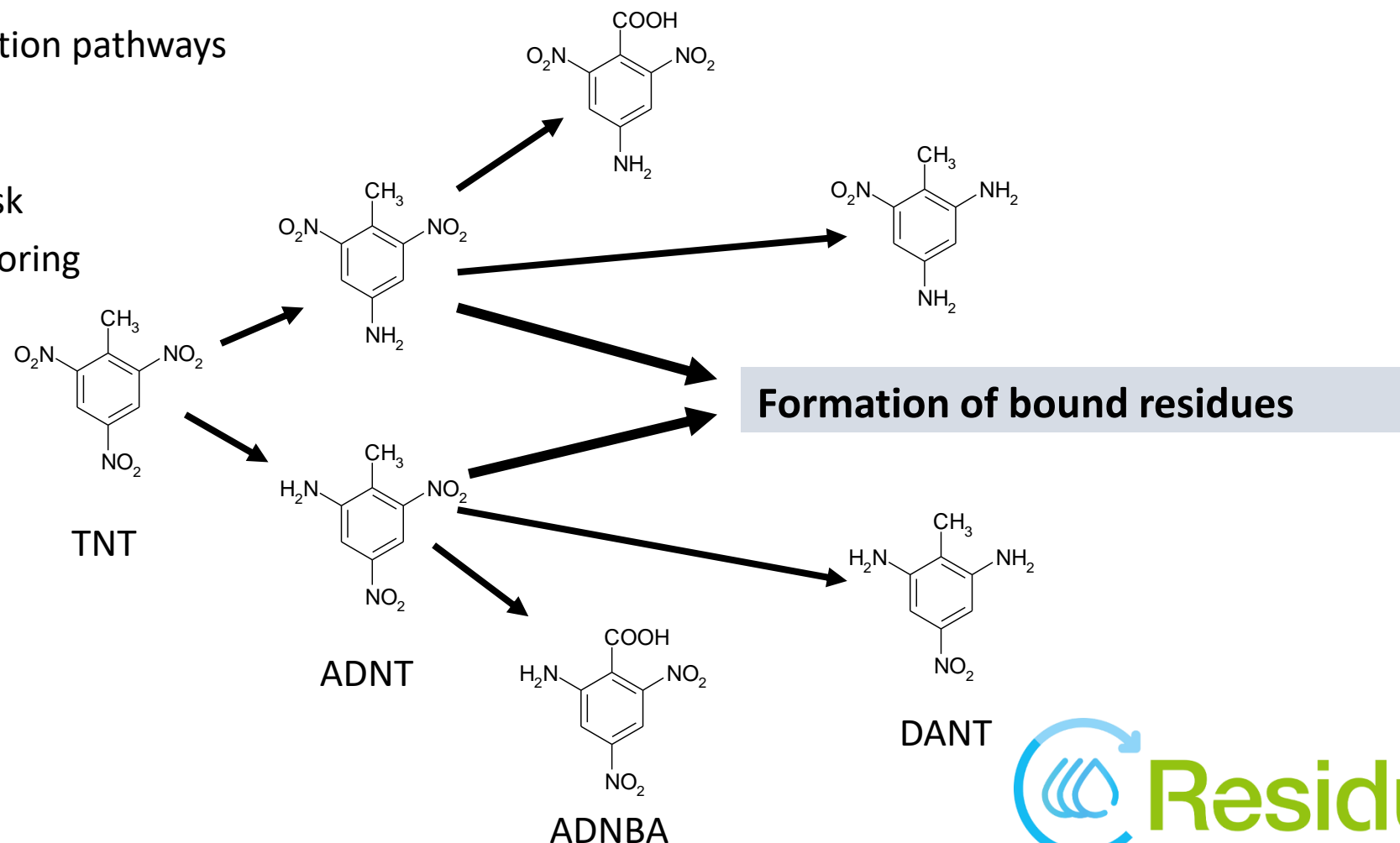
Mass spectrum of a BaP metabolite of radiolabelled (A) or non-labelled (B) test sets



Aim and advantages of radiolabelling



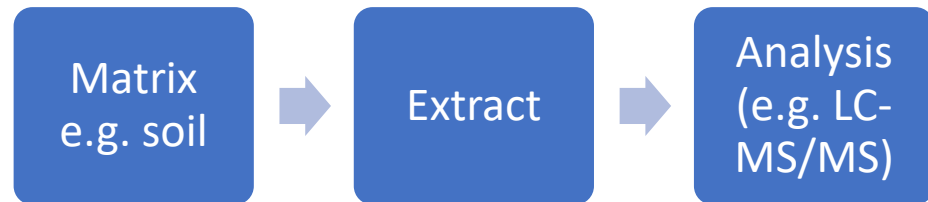
- Elucidation of degradation pathways (route of degradation)
- Basic knowledge for risk assessment and monitoring



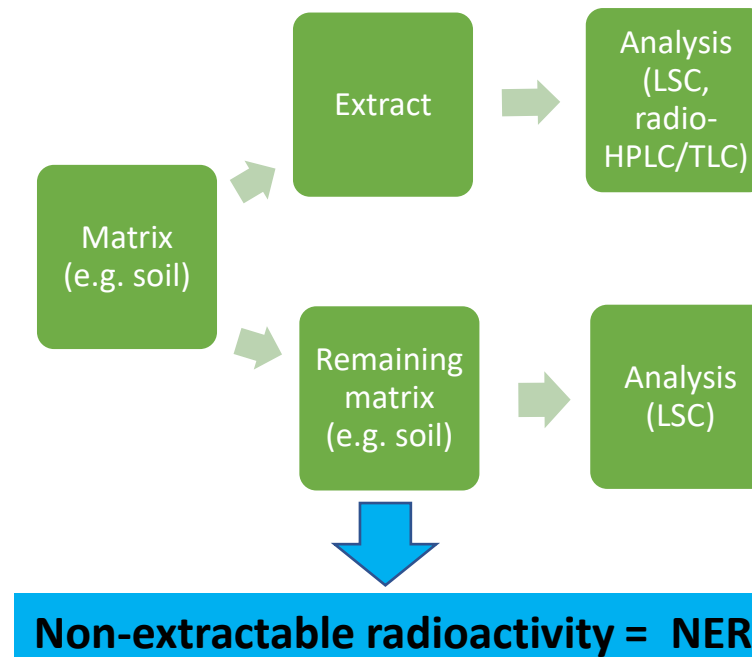
Non-extractable Radioactivity (NER)



Using non-labelled chemicals



Using ^{14}C -labelled chemicals



Mass balances



- Mass balances based on radioactivity recovered in the different compartments are possible

Radioactivity detected in extract(s) [% of applied radioactivity]	Radioactivity detected in the matrix after extraction NER [% of applied radioactivity]	Radioactivity detected in traps for $^{14}\text{CO}_2$ [% of applied radioactivity]	Radioactivity detected in traps for organic volatiles [% of applied radioactivity]	Total recovery [% of applied radioactivity]
90	5	2	n.d.	97
5	3	10	60	78

✓

Test systems



- Test systems for environmental simulation studies in a lab scale - for research or regulatory approval



Investigating the fate of an organic substance

Hydrolysis, Photolysis (OECD 111,

Adsorption/Desorption (OECD 106)

Transformation in Soil (OECD 307)

Transformation in Water/Sediment Systems (OECD 308)

Transformation in Surface Water (OECD 309)

Degradation in Sewage Treatment Plants (OECD 314)

Metabolism in Plants (OECD 501)

Metabolism in Rotational Crops (OECD 502)

Lysimeter Studies



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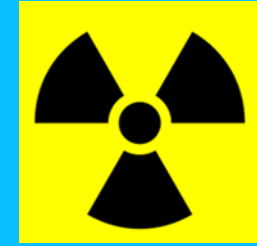
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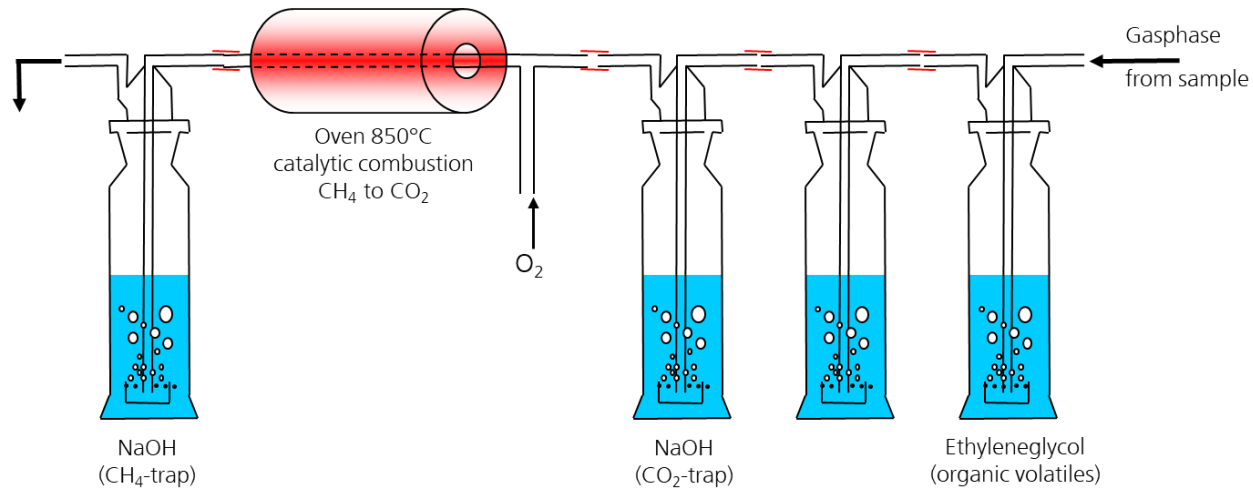


Test systems



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Scheme of gas traps (anaerobic)



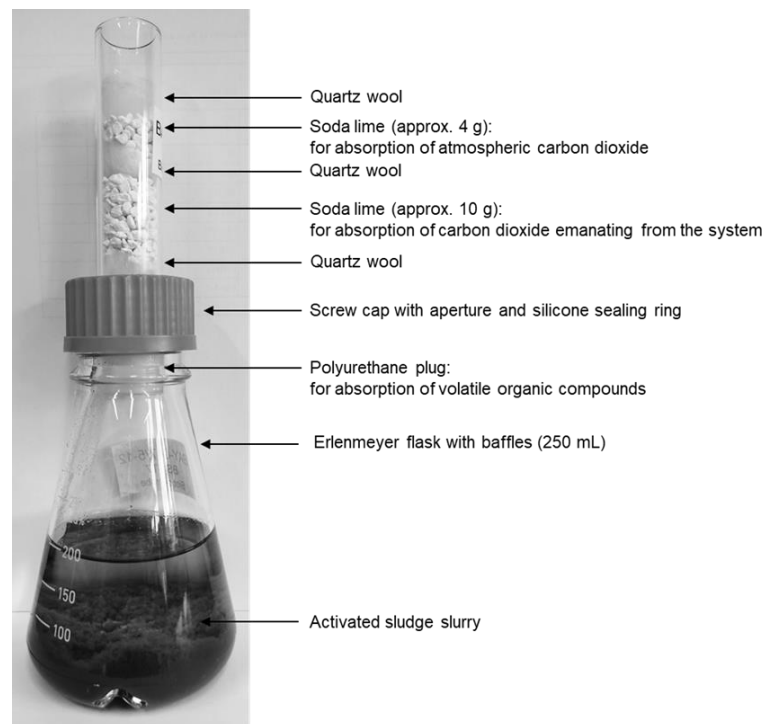
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- Outdoor studies in a bigger scale



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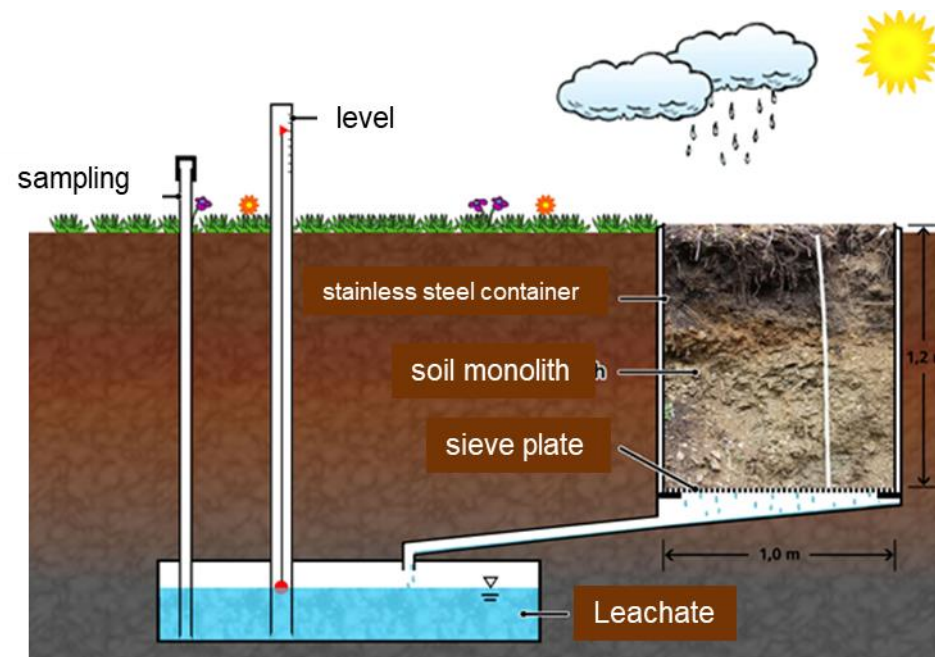
Metabolism in Rotational Crops (OECD 502)

Lysimeter Studies

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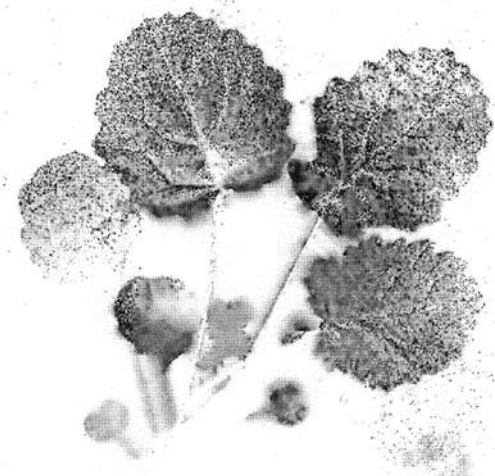
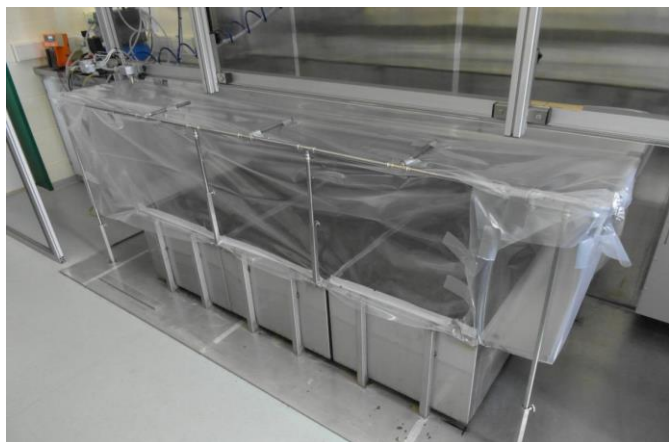
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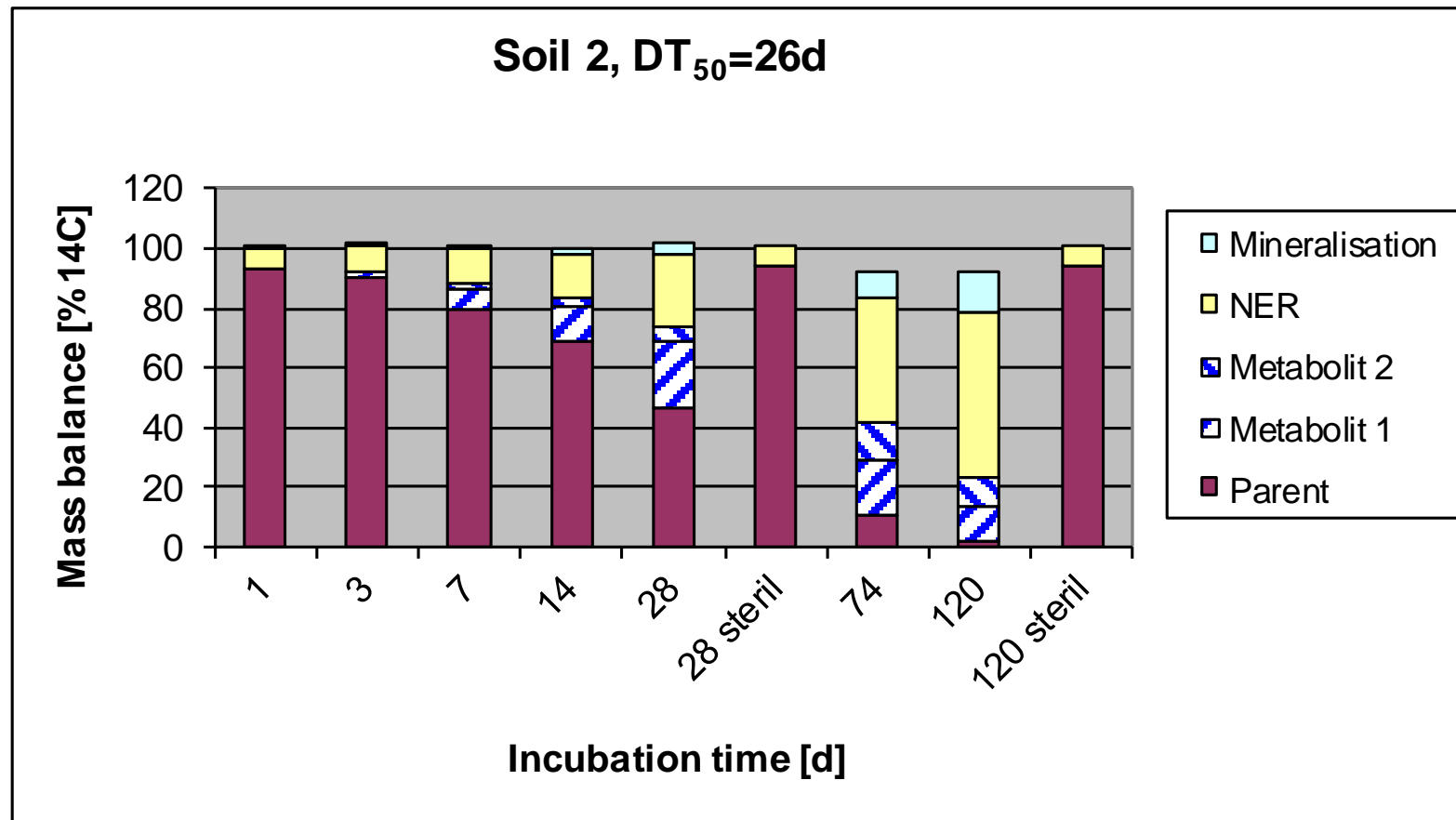
Application



Degradation / Transformation in Soil



- Example 1: Degradation of a pesticide

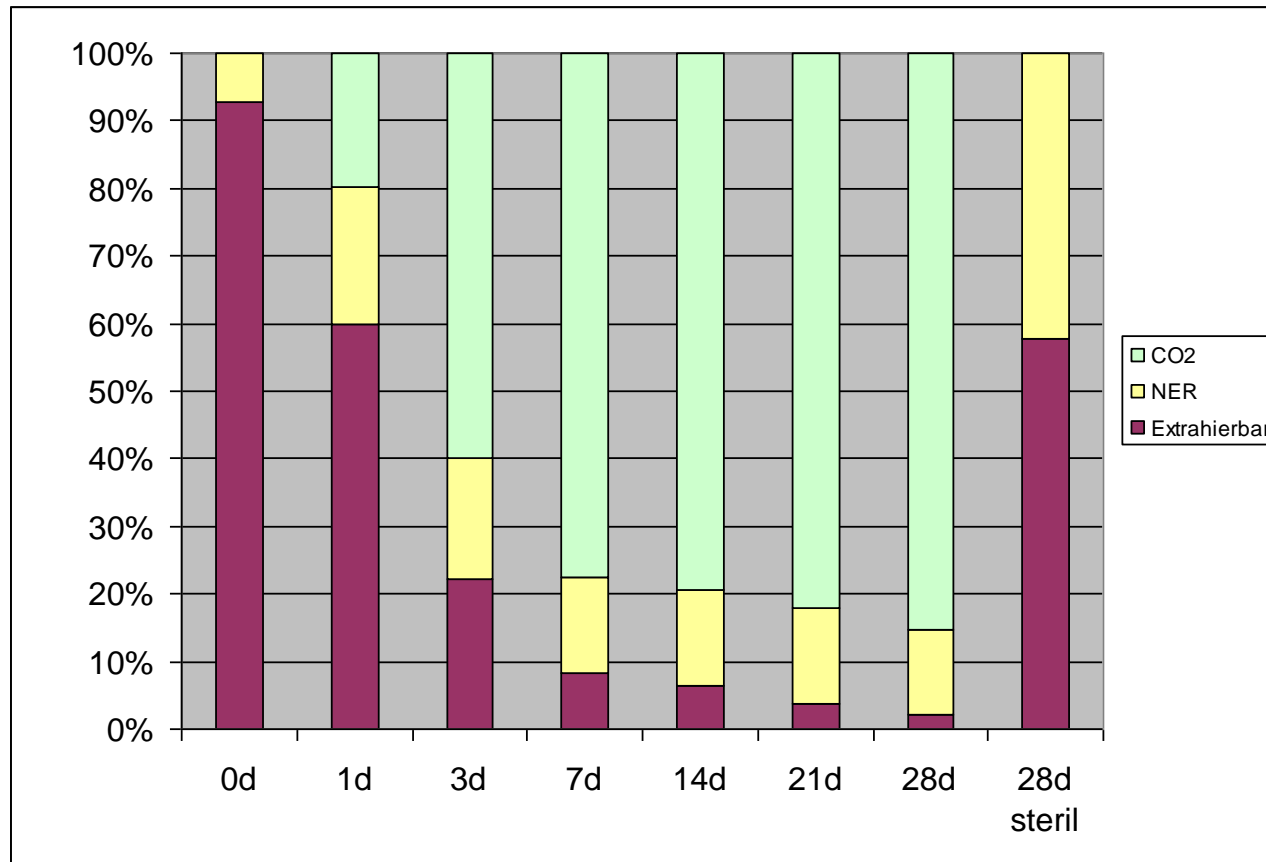


- DT50 of parent compound was 26d, metabolites are formed
- Mineralisation is low
- NER-formation biotic

Degradation / Transformation in Soil



○ Example 2: Degradation of an antibiotic

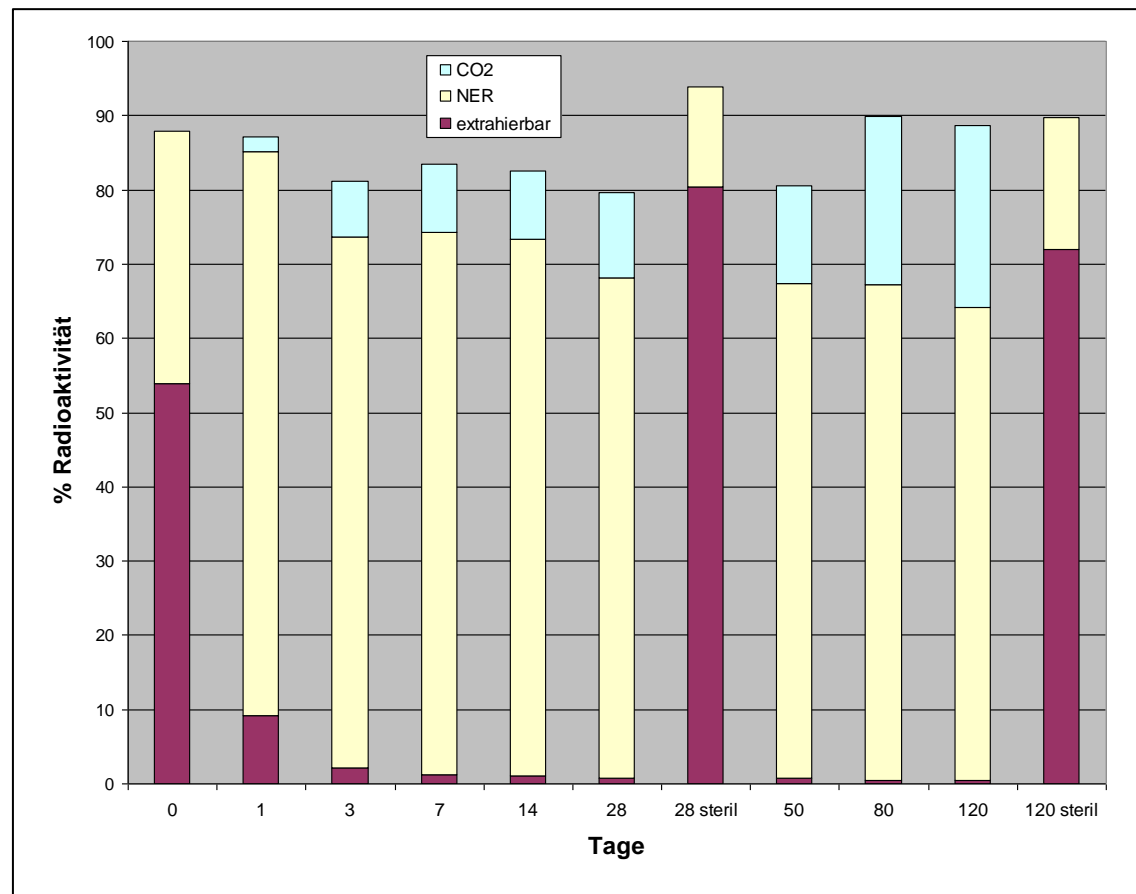


- Rapid mineralisation
- NER-formation abiotic
- Stopped at 28 d (90% Trigger reached)

Degradation / Transformation in Soil



○ Example 3: Degradation of an another antibiotic compound



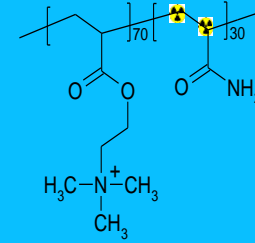
- Rapid NER-generation but:
- Nevertheless considerable mineralisation till study end!

Main reason for NER:
biotic processes!

Apparently bioavailability
of NER

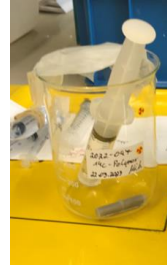
No release of parent
substance!

Degradation of Polymers

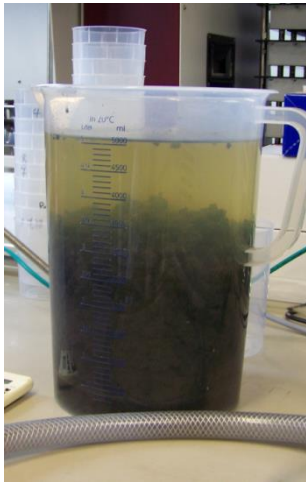


Fate testing of polymeric flocculants

- Synthesis of ¹⁴C-polyacrylamide
Average molecular weight: 6 Mio Da



Sewage sludge



flocculation



dewatering

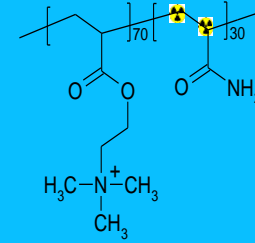


Application to outdoor soil lysimeter



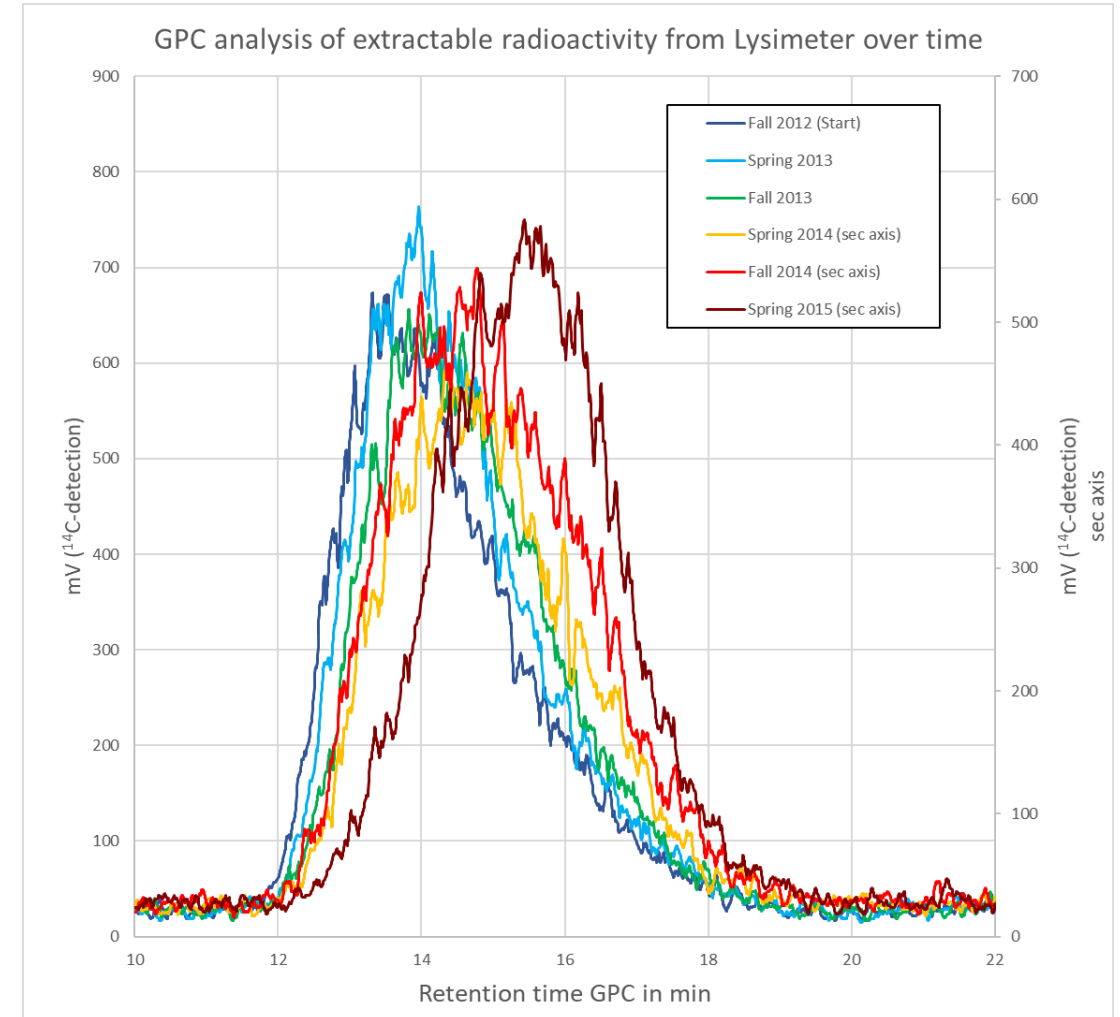
09.18.2008

Degradation of Polymers



Fate testing of polymeric flocculants

- GPC-analysis (separation by hydrodynamic volume of molecules, largest molecules elute first) coupled to ^{14}C -detection
- Over time the distribution of the molecular masses shifted to smaller sizes



RESIDUE



Irrigation of plants with (artificial) wastewater including ^{14}C -labelled Lamotrigine and Carbamazepine



Uptake of the radiolabel into the plants

Many thanks for your attention

Contact

Dieter.hennecke@ime.fraunhofer.de

Kerstin.derz@ime.fraunhofer.de

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